

Magnetism of sodium superoxide

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By combining first-principles electronic-structure calculations with the model Hamiltonian approach, we systematically study the magnetic properties of sodium superoxide (NaO_2), originating from interacting superoxide molecules. We show that NaO_2 exhibits a rich variety of magnetic properties, which are controlled by relative alignment of the superoxide molecules as well as the state of partially filled antibonding molecular π_g -orbitals. The orbital degeneracy and disorder in the high-temperature pyrite phase gives rise to weak isotropic antiferromagnetic (AFM) interactions between the molecules. The transition to the low-temperature marcasite phase lifts the degeneracy, leading to the orbital order and formation of the quasi-one-dimensional AFM spin chains. Both tendencies are consistent with the behavior of experimental magnetic susceptibility data. Furthermore, we evaluate the magnetic transition temperature and type of the long-range magnetic order in the marcasite phase. We argue that this magnetic order depends on the behavior of weak isotropic as well as anisotropic and Dzyaloshinskii-Moriya exchange interactions between the molecules. Finally, we predict the existence of a multiferroic phase, where the inversion symmetry is broken by the long-range magnetic order, giving rise to substantial ferroelectric polarization.

1 Introduction

Sodium superoxide (NaO_2) exhibits a rich variety of properties, which make it an interesting example of multifunctional materials. Being traditionally used as one of the components of oxygen regeneration devices, it emerged recently as a promising material for rechargeable room-temperature batteries.¹

Particularly interesting are the magnetic properties of NaO_2 , which were actively studied in 1970s, together with other alkali superoxides (or hyperoxides).^{2–8} Despite some exoticism, these systems may be regarded as alternative magnetic materials, which are built without traditional transition-metal or rare-earth elements, and where the source of the magnetism is the molecular superoxide complexes O_2^- , arranged in a periodic lattice. Such a behavior is rather unique and related to the Hund's rule effects in the p -electron shell of oxygen molecules, which tend to form a high-spin triplet state.⁹ Since O_2^- is an aspherical object, a particular attention was paid to exploration of principally new effects, associated with the rotational degrees of freedoms of the superoxide molecules. One of such effects is the magnetogyraton – a new type of magnetoelastic coupling between localized magnetic moments and rotations of the molecular groups.^{6,7}

Nevertheless, one very important aspect of molecular complexes was yet overlooked in the early studies of alkali superoxides. In comparison with the regular oxygen molecule, O_2^- acquires an extra electron in the degenerate π_g -shell, which is composed from antibonding molecular orbitals of the p_x - and p_y -type. Thus, the properties of superoxides crucially depend on where this electron will reside. This may depend on several factors. For instance, because of the Jahn-Teller theorem, the degeneracy of the π_g -orbitals should be lifted in the ground state, that typically leads to the long-range orbital ordering. Nevertheless, the direction for lifting the degeneracy depends on the electron-electron interactions, relativistic spin-orbit (SO) interaction, lattice distortions and – in this particular case – the rotations of the superoxide complexes. The magnetic properties of superoxides will replicate the orbital ordering. In the high-temperature phase, however, the Jahn-Teller theorem is no longer applicable and the character of inter-molecular magnetic interactions can be specified by the orbital disorder in a degenerate state. Such orbital effects are well known for the transition-metal oxides, where the active orbitals are the atomic d -orbitals.^{10,11} The new aspect of superoxides is that the same type of the effects can be realized on the molecular π_g -orbitals, which may bring some new functionalities into the canonical problem of magnetic interactions, due to specific geometry and antibonding character of these orbitals. Thus, the recent wave of interest in the alkali superoxides is related to the exploration of the orbital effects and their impact on the magnetic properties.^{12–17} Another important direction is the search of the new type of the ferromagnetic and, possibly, half-metallic materials on the basis of p -

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elements.^{18,19}

In this work, we will present a comprehensive theoretical analysis of the magnetic properties of NaO₂. Amongst alkali superoxides, NaO₂ is one of the most experimentally studied compounds. Particularly, it is the only compound for which details of the lattice distortion and rotations of the superoxide molecules have been reported for the low-temperature marcasite phase.^{3,20} We will show that NaO₂ is an exciting magnetic material, which bears many similarities with more traditional transition-metal oxides. The magnetic properties of the high-temperature pyrite phase are specified by the orbital disorder, which explains isotropic and weakly antiferromagnetic (AFM) character of inter-molecular interactions. On the contrary, the orthorhombic distortion in the low-temperature marcasite phase sets up an orbital order, which results in the quasi-one-dimensional character of inter-molecular interactions. Moreover, the magnetic interactions in this phase are featured by frustration effects. These two factors explain relatively low magnetic transition temperature of NaO₂. Fine details of the magnetic structure are controlled by the SO-related anisotropic and Dzyaloshinskii-Moriya interactions, which lift the degeneracy of the ground state. Moreover, we predict the existence of the multiferroic phase of NaO₂, where the inversion symmetry is broken by the long-range magnetic order due to the frustration.

2 Method

According to the first-principles electronic structure calculations in the local density approximation (LDA), the strongest hybridization in NaO₂ occurs within the superoxide complexes, which leads to the formation of the molecular levels. The characteristic splitting between these levels is several eV. The hybridization between molecules is significantly weaker and leads to the formation of the molecular bands. The characteristic bandwidth is of the order of one eV. Thus, the bands do not overlap with each other and can be identified by using the same notations as for the isolated molecular levels. Typical example of the LDA band structure for the pyrite phase is shown in Figure 1.

LDA is known to be a bad approximation for the description of Coulomb correlations in the narrow-band compounds. For these purposes, it is essential to go beyond LDA. At the same time, we do not need to know all details of the electronic structure: since the magnetic properties of NaO₂ are solely related to the behavior of the π_g -bands, we can concentrate on the description of only this group of states, by constructing an effective low-energy (Hubbard-type) model and including the effect of other states to the definition of parameters of the low-energy model. For these purposes, we construct the Wannier basis for the π_g -band, using the projector-operator technique and isolated molecular orbitals as the trial functions.²²

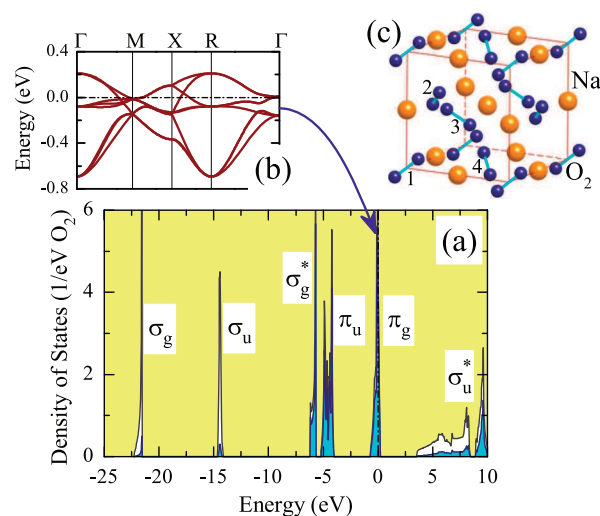


Fig. 1 (a) Total and partial densities of states for the pyrite phase of NaO₂ in the local-density approximation. Shaded areas show contributions of the oxygen 2p states. The positions of the main bands are indicated by symbols. The Fermi level is at zero energy (shown by dot-dashed line). (b) Energy dispersion of the π_g -band, located near the Fermi level. Notations of the high-symmetry points of the Brillouin zone are taken from the book of Bradley and Cracknell.²¹ (c) Crystal structure of the pyrite phase.

Then, matrix elements of LDA Hamiltonian in the Wannier basis give us one-electron part of the model: namely, the site-diagonal elements describe the crystal-field splitting of the molecular π_g -orbitals, while the off-diagonal elements have a meaning of transfer integrals (or the kinetic hoppings between different molecular sites). The effective Coulomb interactions in the π_g -band should take into account the screening by other bands. This screening is calculated by combining constraint-LDA and the random-phase approximation (RPA). This procedure was previously applied to KO₂,¹² which is similar to NaO₂, except for the crystal structure. The details can be also found in the review article.²³

3 Results and Discussions

3.1 Pyrite phase

3.1.1 Structural properties The pyrite phase of NaO₂ is realized above 196 K. It has four superoxide ions in the primitive cell, which are located at $(0, 0, 0)$, $(0, \frac{1}{2}, \frac{1}{2})$, $(\frac{1}{2}, 0, \frac{1}{2})$, and $(\frac{1}{2}, \frac{1}{2}, 0)$, respectively, in units of the cubic lattice parameter a . Each molecule is aligned along one of the four body diagonals of the cube: $[1, \bar{1}, 1]$, $[1, 1, \bar{1}]$, $[\bar{1}, 1, 1]$, or $[1, 1, 1]$. Below 223 K, these molecular axes are ordered as explained in Figure 1c. The corresponding space group is $Pa\bar{3}$. It is believed

that above 223 K, the molecular axes undergo some hindered rotations, leading to an averaged NaCl-type structure with the space group $Fm\bar{3}m$. Nevertheless, the diffuse X-ray scattering indicates that, even above 223 K, the distribution over four possible molecular axes is not entirely random, and the existing between them correlations correspond to the local order of the pyrite type.² Since the physically relevant mechanism, responsible for the inter-molecular exchange interactions in alkali superoxides is the superexchange (SE),¹² which is a local process and depends only on transfer integrals between neighboring molecules,^{10,24} we do need to know all details of the orientational disorder and assume the ordered $Pa\bar{3}$ structure in all temperature range above 196 K. We use experimental parameters of the crystal structure of NaO₂, reported in ref. 3.

3.1.2 Parameters of electronic model The electronic structure of the pyrite phase of NaO₂ in LDA is explained in Figure 1. As was discussed in Section 2, our first goal is the construction of an effective Hubbard-type model for the narrow oxygen π_g -band, located near the Fermi level. Due to the high $Pa\bar{3}$ symmetry, the molecular π_g -orbitals remain degenerate and there is no crystal-field splitting. Therefore, the one-electron part of the model will include only the transfer integrals, connecting different O₂⁻ sites. For each pair of the O₂⁻ molecules, located at i and j , the transfer integrals between molecular π_g -orbitals are the 2×2 real matrices \hat{t}_{ij} . The form of these matrices depends on the relative position of the sites i and j , and the choice of the local coordinate frame at each of these sites.²⁵ However, for the purposes of this section, it is sufficient to know only the averaged transfer integrals $\bar{t}_{ij} \equiv \|\hat{t}_{ij}\|_F$, expressed in terms of Frobenius matrix norm $\|\hat{t}_{ij}\|_F = \sqrt{\text{Tr}(\hat{t}_{ij}\hat{t}_{ij}^T)}$, where Tr is the trace over orbital indices. For the nearest-neighbor (NN) sites, all \bar{t}_{ij} are the same ($\bar{t}_{ij} \equiv \bar{t}$) and do not depend on the coordinate frame. Using numerical values of transfer integrals \hat{t}_{ij} , \bar{t} can be evaluated as $\bar{t} = 68$ meV. As a test, the bandwidth for the face-centered cubic lattice can be estimated as $12\bar{t} = 816$ meV, which is well consistent with results of LDA calculations for the π_g -band (Figure 1).

For the $Pa\bar{3}$ symmetry, the screened intra-molecular interactions between π_g -electrons can be described in terms of two independent Kanamori parameters:²⁶ the intra-orbital Coulomb repulsion U and the intra-molecular (Hund's) exchange J_H . The third parameter, the so-called inter-orbital Coulomb repulsion U' , is related with the former two by the identity $U' = U - 2J_H$. The direct calculations, using the combined constraint-LDA plus RPA approach for the screening,^{12,23} yield $U = 3.65$ eV and $J_H = 0.61$ eV.

3.1.3 Spin Hamiltonian and magnetic properties Now, we are ready to construct the isotropic (Heisenberg-type) spin

Hamiltonian,

$$\mathcal{H}_H = - \sum_{i>j} J_{ij} \mathbf{S}_i \mathbf{S}_j, \quad (1)$$

and evaluate parameters of this Hamiltonian for the pyrite phase, using the theory of SE interactions.^{10,24} In these notations, $S = 1/2$ and the summation runs over inequivalent pairs of molecular sites. The SE theory is basically the second order perturbation theory with respect to the transfer integrals in the bond i - j , where one should evaluate the energy gain (\mathcal{T}_{ij}) caused by the virtual hoppings from the space of occupied states at the site i to the subspace of unoccupied states at the site j (and vice versa) and to do so for the ferromagnetic (FM, $\uparrow\uparrow$) and AFM ($\uparrow\downarrow$) configurations of spins. Then, the exchange coupling J_{ij} is obtained by mapping these energies onto the spin Hamiltonian (1), which yields $J_{ij} = (\mathcal{T}_{ij}^{\downarrow\downarrow} - \mathcal{T}_{ij}^{\uparrow\uparrow})/(2S^2)$.

In the O₂⁻ molecule, there are three electrons, residing on four molecular spin-orbitals of the π_g -symmetry. Therefore, there will be two occupied orbitals with the majority (\uparrow) spin and one occupied orbital with the minority (\downarrow) spin. Very generally, the latter (electronic) orbital can be presented in the form:⁹

$$|\psi_i^e\rangle = \cos \frac{\vartheta_i}{2} e^{i\varphi/2} |p_x\rangle + \sin \frac{\vartheta_i}{2} e^{-i\varphi/2} |p_y\rangle,$$

in terms of two π_g -orbitals of the p_x and p_y symmetry (where the direction of z is taken along the molecular axis, $0 \leq \vartheta_i \leq \pi$, and $0 \leq \varphi_i \leq 2\pi$). Then, the unoccupied (hole) orbital should be orthogonal to $|\psi_i^e\rangle$:

$$|\psi_i^h\rangle = -\sin \frac{\vartheta_i}{2} e^{i\varphi/2} |p_x\rangle + \cos \frac{\vartheta_i}{2} e^{-i\varphi/2} |p_y\rangle.$$

In the absence of the crystal-field splitting, the angles ϑ_i and φ_i can take arbitrary values. In the ordered state, they should minimize the total energy of the crystal.¹⁰ Nevertheless, the pyrite phase exists only in the high temperature state, where there should be no spin or orbital order. Therefore, in order to consider spin interactions, J_{ij} should be averaged over ϑ_i and φ_i , that would simulate the effect of the orbital disorder.¹⁰

Let us first evaluate $\mathcal{T}_{ij}^{\uparrow\uparrow}$. Since the hoppings are permitted only between orbitals with the same spin, we will have two such contributions, corresponding to the transitions from $|\psi_i^e\rangle$ to $|\psi_j^h\rangle$ and from $|\psi_j^e\rangle$ to $|\psi_i^h\rangle$. Therefore, $\mathcal{T}_{ij}^{\uparrow\uparrow}$ will be given by the following expression:

$$\mathcal{T}_{ij}^{\uparrow\uparrow} = - \frac{|\langle \psi_i^e | \hat{t}_{ij} | \psi_j^h \rangle|^2 + (i \leftrightarrow j)}{U - 3J_H},$$

where the denominator is the energy cost for transferring an electron between sites i and j in the ionic limit (the inter-orbital Coulomb repulsion $U' = U - 2J_H$ minus the exchange interaction J_H in the case of the $\uparrow\uparrow$ configuration of spins). For the $\uparrow\downarrow$ configuration, both states $|\psi_i^e\rangle$ and $|\psi_i^h\rangle$ with the same

spin as $|\psi_j^h\rangle$ will be occupied. Therefore, $\mathcal{T}_{ij}^{\uparrow\downarrow}$ will have four contributions:

$$\mathcal{T}_{ij}^{\uparrow\downarrow} = -\frac{|\langle\psi_i^h|\hat{t}_{ij}|\psi_j^h\rangle|^2 + (i \leftrightarrow j)}{U} - \frac{|\langle\psi_i^e|\hat{t}_{ij}|\psi_j^h\rangle|^2 + (i \leftrightarrow j)}{U - 2J_H}.$$

The energy cost for the first process is U (after transferring an electron from the site i , two remaining electrons occupy the same type of orbitals) and, for the second process, it is $U' = U - 2J_H$ (remaining electrons occupy orbitals of different type). Note that there is no exchange interaction J_H between states with different spins.

Then, we average $\mathcal{T}_{ij}^{\uparrow\uparrow}$ and $\mathcal{T}_{ij}^{\uparrow\downarrow}$ over the angles ϑ_i , φ_i , ϑ_j , and φ_j , and treat these angles as independent parameters. This yields $|\langle\psi_i^h|\hat{t}_{ij}|\psi_j^h\rangle|^2 = |\langle\psi_i^e|\hat{t}_{ij}|\psi_j^h\rangle|^2 = \bar{t}^2/4$.²⁷ Then, the NN exchange coupling can be estimated as

$$J = \bar{t}^2 \left(\frac{1}{U - 3J_H} - \frac{1}{U - 2J_H} - \frac{1}{U} \right).$$

Thus, the sign of J is solely determined by the ratio U/J_H , and for $U > (3 + \sqrt{3})J_H$ (which is satisfied for the actual values of U and J_H), the NN interaction is AFM. Using numerical values of U , J_H , and \bar{t} , reported above, J can be estimated as $J = -0.63$ meV. Then, the Curie-Weiss temperature $\theta_{CW} = zJS(S+1)/3k_B$ ($z = 12$ being the coordination number) is $\theta_{CW} = -22$ K, which is consistent with the experimental temperature of -31 K.²

We can apply the same strategy to the high-temperature body-centered tetragonal phase of KO_2 , using parameters of electronic model reported in ref. 12. In this case, two strongest magnetic interactions between first and second NN (located along the tetragonal a -axis and the body diagonal of the tetragonal cell) can be estimated as -1.38 and -3.00 meV, respectively. This yields $\theta_{CW} = -87$ K, which is in reasonable agreement with the experimental value of -44 K.² This model analysis naturally explains results of numerical calculations, reported in ref. 12.²⁸

Thus, the negative value of θ_{CW} is the generic feature of alkali superoxides in the high-temperature regime,^{2,29,30} which can be attributed to the orbital disorder. The positive value of $\theta_{CW} = 33$ K, observed in the narrow temperature range ($196 \text{ K} < T < 223 \text{ K}$) of NaO_2 , may be related to the onset of the spin-orbital coupling.³¹ Nevertheless, in the higher temperature range, the spin and orbital degrees of freedom become independent from each other, that eventually leads to the negative value of θ_{CW} .

3.2 Marcasite phase

3.2.1 Structural properties The marcasite phase of NaO_2 is realized below 196 K. It crystallizes in the orthorhombic $Pnmm$ structure, containing two superoxide ions

in the primitive cell, which are located at $(0,0,0)$ and $(\frac{1}{2}, \frac{1}{2}, \frac{1}{2})$, respectively, in units of orthorhombic translations \mathbf{a} , \mathbf{b} , and \mathbf{c} (see Figure 2c). These ions can be transformed to each other by the glide reflections $\{\hat{m}_{\mathbf{a}}|\mathbf{a}/2+\mathbf{b}/2+\mathbf{c}/2\}$ and $\{\hat{m}_{\mathbf{b}}|\mathbf{a}/2+\mathbf{b}/2+\mathbf{c}/2\}$, where $\hat{m}_{\mathbf{a}(\mathbf{b})}$ is the mirror reflection of the orthorhombic $\mathbf{a}(\mathbf{b})$ axis, and $(\mathbf{a}/2+\mathbf{b}/2+\mathbf{c}/2)$ is the translation along the body diagonal of the orthorhombic cell.^{2,3} The magnetic order is realized below 43 K. However, its type

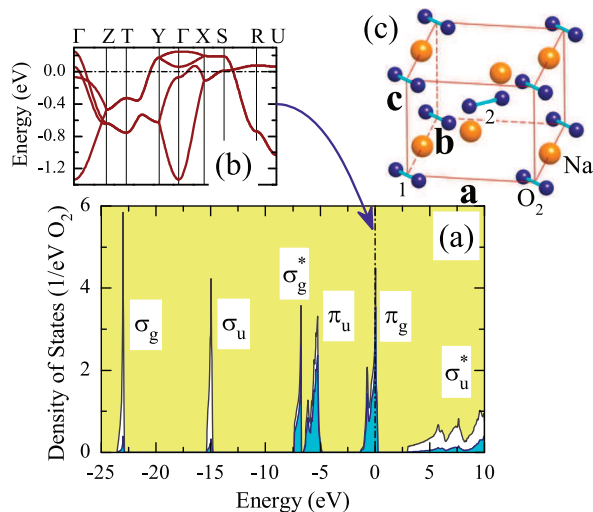


Fig. 2 (a) Total and partial densities of states for the marcasite phase of NaO_2 in the local-density approximation. Shaded areas show contributions of the oxygen $2p$ states. The positions of the main bands are indicated by symbols. The Fermi level is at zero energy (shown by dot-dashed line). (b) Energy dispersion of the π_g -band, located near the Fermi level. Notations of the high-symmetry points of the Brillouin zone are taken from the book of Bradley and Cracknell.²¹ (c) Crystal structure of the marcasite phase.

is unknown. In all the calculations we use the experimental parameters of the crystal structure at $T = 77.3$ K, reported in ref. 3.

3.2.2 Parameters of electronic model The electronic structure of the marcasite phase of NaO_2 in LDA is shown in Figure 2. Again, our first goal is the construction of the effective Hubbard-type model for the oxygen π_g -band, located near the Fermi level. The new aspect of the marcasite structure is that the local symmetry is low. The point group includes only the inversion \hat{I} , the mirror reflection $\hat{m}_{\mathbf{c}}$, and the 180° rotation around the orthorhombic \mathbf{c} -axis ($\hat{C}_2^{\mathbf{c}}$). Therefore, the π_g -levels belong to different representations and will generally split. More specifically, one of π_g -orbitals lies in the \mathbf{ab} -plane (we will call it the p_{xy} -orbital, which is a linear combination of the p_x - and p_y -orbitals) and another orbital is parallel to the \mathbf{c} -axis (the p_z -orbital). We adopt the orbital indices, where $m =$

1 and 2 correspond to p_{xy} and p_z , respectively. These orbitals are split by the crystal field and the value of this splitting for the $T = 77$ K structure is about 213 meV. Moreover, the p_z -orbitals are located higher in energy and periodically ordered, as explained in Figure 3. This orbital ordering has profound consequences on the behavior of inter-molecular magnetic interactions,¹⁰ which will be discussed below.

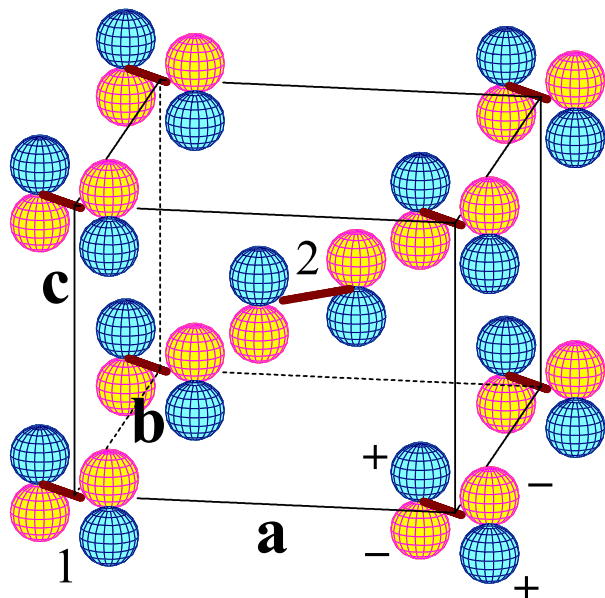


Fig. 3 Distribution of unoccupied antibonding molecular orbitals of the p_z -symmetry (the so-called orbital ordering) in the marcasite phase of NaO_2 . Positive and negative lobes of the p_z -orbitals are shown by different colors. Two sublattices of superoxide ions are denoted by indices '1' and '2'.

The behavior of the transfer integrals in the basis of the p_{xy} - and p_z -orbitals is explained in Figure 4. The strongest transfer occurs between p_z -orbitals along the orthorhombic c -axis. Nevertheless, this is to be expected from the form of the neighboring p_z -orbitals, which are directed towards each other. As we will see below, these transfer integrals are responsible for the strong AFM coupling and formation of the quasi-one-dimensional AFM chains along c . The second strongest transfer occurs between the p_{xy} -orbitals along the a -axis. Nevertheless, in the marcasite phase, all p_{xy} -orbitals are occupied and, therefore, transfer integrals between them will not contribute to the SE interactions. Somewhat strong transfer occurs between NN sites of the sublattices 1 and 2 along the body diagonals. Other transfer integrals are small. Nevertheless, as we will see below, they play an important role in the formation of the long-range magnetic order at finite T .

The values of screened intra-orbital Coulomb repulsion U and the exchange interaction J_H are 3.68 and 0.61 eV, respectively. In principle, due to the low symmetry, the screening

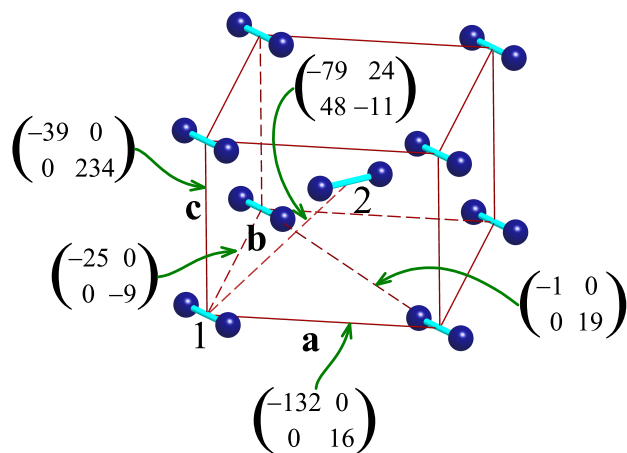


Fig. 4 Matrices of transfer integrals $t_{ij}^{mm'}$ (in meV), associated with different bonds in the marcasite phase of NaO_2 . For each matrix, $m(m') = 1$ corresponds to the p_{xy} type of orbitals, and $m(m') = 2$ corresponds to the p_z type of orbitals. Two sublattices of superoxide ions are denoted by indices '1' and '2'.

of U will be slightly different for the orbitals of the p_{xy} - and p_z -type. However, this difference is only about 1% and can be neglected.

3.2.3 Parameters of spin Hamiltonian Since all transfer integrals are small in comparison with the Coulomb repulsion, the parameters of spin Hamiltonian (1) can be also evaluated using the SE theory, and considering all types virtual hoppings into the subspace of unoccupied p_z -states.^{10,24} The behavior of obtained parameters is explained in Figure 5. Within each

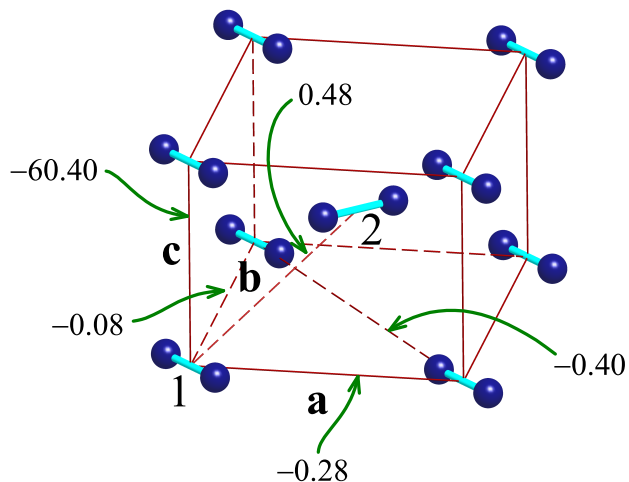


Fig. 5 Isotropic exchange interactions (in meV), associated with different bonds in the marcasite phase of NaO_2 . Two sublattices of superoxide ions are denoted by indices '1' and '2'.

magnetic sublattice, all transfer integrals are diagonal with respect to the orbital indices. Then, since all p_{xy} orbitals are occupied, only p_z -orbitals will contribute to the SE interactions. For the sublattice 1, the transfer integrals between these p_z -orbitals are (in meV): $t_a = 16$, $t_b = -9$, $t_c = 234$, and $t_{a-b} = 19$, operating along to the orthorhombic axes **a**, **b**, and **c**, and the face diagonal **a-b**, respectively (see Figure 4). The transfer integrals t_{a+b} along another face diagonal are small. This seems to be natural. Along **a-b**, the molecular axes are oriented towards each other, leading to larger overlap between the molecular orbitals. Furthermore, the **a**-axis is about 20 % shorter than **b**, and the molecular axes are additionally canted towards **a**. This will lead to a larger overlap along **a**. Nevertheless, we would like to emphasize that all these effects are sensitive to the details of the crystal structure and the additional rotations of the O_2^- molecules may change the situation. Some effects of these rotations will be discussed in Section 3.2.5.

For the sublattice 2, one should interchange t_{a-b} and t_{a+b} .

Corresponding parameters of isotropic exchange interactions for $S = 1/2$ can be obtained using the formula $J_R = -4t_R^2/U$. Thus, J_R in Figure 5 simply reflects the behavior of transfer integrals, depicted in Figure 4. The transfer integrals between different sublattices have large off-diagonal elements, which will contribute to both FM and AFM coupling. These contributions are inversely proportional to $(U - 3J_H)$ and $(U - 2J_H)$, respectively. Because of J_H , the FM contributions will prevail. On the other hand, the transfer integrals between p_z -orbitals, which contribute only to the AFM coupling, are small. This explains the FM character of NN interactions between different sublattices $J_{12} = 0.48$ meV.

The isotropic exchange interactions are frustrated. The largest interaction J_c will tend to align the spins in each sublattice antiferromagnetically. Then, each superoxide ion will form an equal number of FM and AFM bonds with the NN ions of other sublattice. In such a situation, the spins of different sublattices can arbitrarily rotate relative to each other and the ground state will be infinitely degenerate. The inter-sublattice coupling J_{12} will lift this degeneracy in favor of a classically ordered spin-spiral state. However, since $|J_c| \gg J_{12}$, the effect is small and can be neglected. We believe that a more realistic scenario for lifting the degeneracy is the exchange striction and (or) anisotropic exchange interactions. The latter will be considered in Section 3.2.6.

3.2.4 Magnetic susceptibility In this section, we calculate the magnetic susceptibility (χ) of the spin model (1), using realistic parameters derived above. For these purposes we employ the quantum Monte Carlo loop algorithm,³² as implemented in the ALPS simulation package.^{33,34} The calculations have been performed using periodic boundary conditions for the superlattice including 60 unit cells along the c-axis, and 2 unit cells along the **a**- and **b**-axes.

Results of these calculations are explained in Figure 6. The

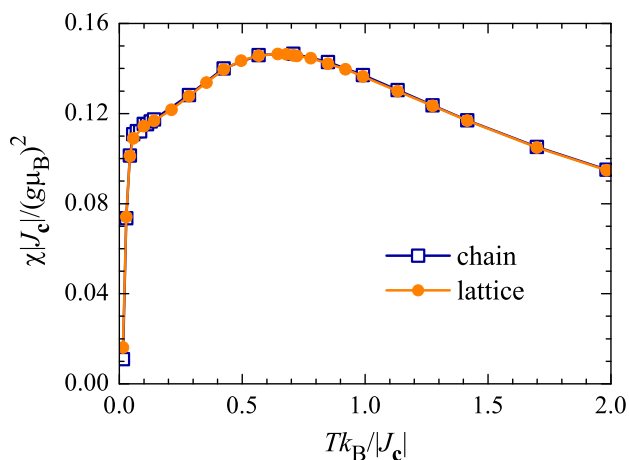


Fig. 6 Magnetic susceptibility (per one superoxide ion), calculated for the isolated antiferromagnetic chain with the nearest-neighbor exchange interaction J_c and the three-dimensional lattice of weakly interacting chains.

shape of χ is mainly controlled by the intrachain interaction J_c . Since other interactions are small, the magnetic susceptibility is practically indistinguishable from that of the isolated AFM chain with the leading NN coupling J_c . The latter was discussed in details by Bonner and Fisher in their canonical work.³⁵ Namely, χ has a broad maximum at $T_{\max} \approx 0.64|J_c|/k_B$. For $|J_c| = 60.4$ meV (see Figure 5), it should correspond to $T_{\max} \approx 550$ K. Thus, in the region $43 \text{ K} < T < 196 \text{ K}$, the magnetic susceptibility is expected to increase. This finding is consistent with the experimental data.² It also supports earlier considerations,^{4,5} based on the semi-empirical model analysis. However, the absolute value of susceptibility at the maximum, $\chi_{\max} \approx 0.15(g\mu_B)^2/|J_c|$,³⁵ can be estimated as $\chi_{\max} \approx 3 \cdot 10^{-4} \text{ cm}^3/\text{mol}$ (using $g = 2$ for the Landé g -factor). This is about 3-5 times smaller than the experimental value of susceptibility.² Naively, one would conclude that such a discrepancy is caused by the overestimation of $|J_c|$ (for example, due to the underestimation of the Coulomb repulsion U). However, the situation is not so simple: if $|J_c|$ were smaller, T_{\max} would be also smaller and we would have serious difficulties in explaining nearly linear experimental dependence of χ on T in the temperature interval $43 \text{ K} < T < 196 \text{ K}$. The discrepancy can be partly resolved by using larger value of $g \approx 2.05 \div 2.10$, due to partial unquenching of the local orbital moment. However, this would increase theoretical χ only by less than 10%. Thus, the discrepancy persists. Apparently, the experimental data themselves may be influenced by the defects and other extrinsic effects. Note that the marcasite phase exists only in the finite temperature range. In such a situation, it is not easy to find the asymptotic behavior and sub-

tract, for instance, the uniform contribution to χ , which may be affected by the sample surroundings.³⁰ We, therefore, propose to carefully reexamine the magnetic susceptibility data. Otherwise, the quasi-one-dimensional picture for the behavior of magnetic susceptibility cannot explain simultaneously the linear character and the absolute value of χ .

3.2.5 Magnetic transition temperature As was pointed out above, due to frustrations, the isotropic exchange interactions alone cannot stabilize the magnetic order between different sublattices. Nevertheless, they do stabilize the magnetic order in each of the sublattices, although the situation is also not simple and severely hampered by quasi-one-dimensional character of the problem. If one considers only the strongest AFM interactions within the chain, according to the Mermin-Wagner theorem,³⁶ there would be no long range magnetic order at finite T . Therefore, the long-range order, if exists, is stabilized by weak interchain interactions. In this section, we evaluate the transition temperature of this magnetic order.

First, from the mean-field approximation for the Heisenberg model (1), we find that the classical ground state corresponds to the propagation vector $\mathbf{Q} = (\pi/a, 0, \pi/c)$. Thus, the magnetic structure is AFM along the orthorhombic \mathbf{a} - and \mathbf{c} -axes, and FM along the \mathbf{b} -axis. It is favored by relatively strong interactions $J_{\mathbf{a}}$ and $J_{\mathbf{a-b}}$ (for the sublattice 1), and is disfavored by the weak interaction $J_{\mathbf{b}}$ (see Figure 5). Then, the Néel temperature can be found from the expression:^{37,38}

$$k_{\text{B}}T_{\text{N}} = z_{\perp}|J_{\perp}|A \ln^{1/2} \left(\frac{\Lambda|J_{\text{c}}|}{k_{\text{B}}T_{\text{N}}} \right),$$

which combines exact results for the analytically solvable model in one dimension with the mean-field treatment for the interchain interactions.³⁷ The numerical calculations for the single chain yield $A \approx 0.32$ and $\Lambda \approx 5.8$.³⁹ In the original formulation,^{37,38} J_{\perp} is an effective interchain exchange coupling and z_{\perp} is the coordination number in the \mathbf{ab} -plane. Therefore, for our purposes, $z_{\perp}|J_{\perp}|$ should be replaced by $2(|J_{\mathbf{a}}| + |J_{\mathbf{a-b}}| - |J_{\mathbf{b}}|) = 1.2$ meV, which yields $T_{\text{N}} \approx 11$ K. Note that, due to the frustration of magnetic interactions, there will be no mean field acting on the spins of the sublattice 1 from the sublattice 2 (and vice versa).

The mean-field theories generally overestimate T_{N} . Therefore, as an alternative possibility, we also considered the random-phase approximation (RPA),⁴⁰

$$\frac{1}{k_{\text{B}}T_{\text{N}}} = \sum_{\mathbf{q}} \frac{4}{J(\mathbf{Q}) - J(\mathbf{q})}, \quad (2)$$

which takes into account collective excitations and is expected to provide a better description for the interchain coupling.³⁸ In this case, $J(\mathbf{q}) = \sum_{\mathbf{R}} J_{\mathbf{R}} e^{i\mathbf{q}\cdot\mathbf{R}}$ is the Fourier image of $\{J_{\mathbf{R}}\}$. Then, Eq. (2) yields $T_{\text{N}} \approx 6$ K, which is consistent with the above mean-field estimate.

According to the experimental data,² the magnetic order (of unknown type) is believed to occur below 43 K, which is substantially higher than our estimate of T_{N} . However, it should be noted that the value of T_{N} is controlled by very small magnetic interactions between the chains, which may be sensitive to the rotations of the O_2^- molecules. Since we used the experimental structure parameters measured at $T = 77$ K (i.e., far above T_{N}), it is possible that the additional rotations of O_2^- below T_{N} may change the situation. In order to check this possibility, we have performed additional calculations, using the experimental structure parameters at $T = 18$ K, which were recently reported in ref. 20. We have found that J_{c} does not change too much (the new value is $J_{\text{c}} = -59$ meV). Other exchange interactions are also relatively small. However, they undergo some changes, which do affect T_{N} . Namely, the strongest interchain interactions for the $T = 18$ K structure are $J_{\mathbf{a-b}} = -0.61$ meV and $J_{\mathbf{a+c}} = -0.63$ meV. They stabilize the classical ground state with $\mathbf{Q} = (0, \pi/b, \pi/c)$. The corresponding magnetic transition temperature can be estimated in the RPA as $T_{\text{N}} \approx 50$ K. Thus, the discrepancy with the experimental data can be resolved by taking into account the additional rotations of the O_2^- molecules below T_{N} .

3.2.6 Anisotropic interactions In this section, we consider anisotropic exchange interactions, which are caused by the relativistic SO coupling, namely, the anisotropic symmetric interactions:

$$\mathcal{H}_{\text{AS}} = \sum_{i>j} \mathbf{S}_i \hat{\tau}_{ij} \mathbf{S}_j,$$

and the antisymmetric Dzyaloshinskii-Moriya (DM) interactions:^{41,42}

$$\mathcal{H}_{\text{DM}} = \sum_{i>j} \mathbf{d}_{ij} [\mathbf{S}_i \times \mathbf{S}_j].$$

Both types of interactions operate only between different molecular sites, and due to the Kramers theorem for the spin 1/2, there will be no single-ion anisotropy. The parameters $\hat{\tau}_{ij}$ and \mathbf{d}_{ij} can be also obtained by considering the effects of the SO coupling in the framework of the SE theory.⁴³ These interactions are relatively weak. Nevertheless, they play a very important role in lifting the degeneracy of the magnetic ground state and stabilizing the inter-sublattice magnetic order. The effect of these interactions can be rationalized in the following way.

The main contribution to \mathcal{H}_{AS} comes from the NN interactions along the \mathbf{c} -axis within sublattices. Corresponding tensor $\hat{\tau}_{\mathbf{c}}$ for the sublattice 1 has the following form (in meV):

$$\hat{\tau}_{\mathbf{c}} = \begin{pmatrix} -0.04 & 0.47 & 0 \\ 0.47 & -0.04 & 0 \\ 0 & 0 & 0.08 \end{pmatrix}.$$

This means that the easy magnetization direction lies in the \mathbf{ab} -plane and is perpendicular to the molecular axes. As expected, $\hat{\tau}_{\mathbf{c}}$ is invariant under the mirror reflection $\hat{m}_{\mathbf{c}}$, which

transform the sublattice 1 to itself. Similar tensor for the sublattice 2 can be obtained by the mirror reflections \hat{m}_a or \hat{m}_b , which, in the combination with the translations $(\mathbf{a}+\mathbf{b}+\mathbf{c})/2$, transform the sublattice 1 to the sublattice 2. Other anisotropic symmetric interactions are small and can be neglected. Moreover, since superoxide ions are located in the inversion centers, there will be no DM interactions operating within each of sublattices. Thus, the anisotropic symmetric interactions will form a noncollinear magnetic structure where all spins lie in the **ab**-plane, perpendicular to their molecular axes.

The DM interactions operate between different sublattices. For example, in the bond 1-2 (see Figure 5), the DM vector is $\mathbf{d}_{12} = (0.06, -0.01, 0.01)$ meV. The DM parameters in other NN bonds can be obtained by applying the symmetry operations of the space group $Pnmm$.⁴⁴ Then, in the classical picture, every spin will experience a force $\mathbf{f}_i = -\partial \mathcal{H}_{DM} / \partial \mathbf{S}_i = \sum_j [\mathbf{d}_{ij} \times \mathbf{S}_j]$ from its neighboring sites. Since all spins $\{\mathbf{S}_j\}$ are confined in the **ab**-plane and form the $\mathbf{Q} = (\pi/a, 0, \pi/c)$ AFM structure, it is easy to show that \mathbf{f}_i should be parallel to the **c**-axis. Thus, the DM interactions will lead to a small canting of spins off the **ab**-plane. Moreover, since $\mathbf{d}_{21} = -\mathbf{d}_{12}$, \mathbf{f}_i will have opposite directions at different molecular sites of the unit cell. Therefore, we expect no weak ferromagnetism in this case.^{41,42}

3.2.7 Multiferroic phase The phenomenon of multiferroicity, or the coupling between magnetism and ferroelectricity, was intensively studied in transition-metal oxides.^{45–47} The key moment for obtaining the spontaneous ferroelectric (FE) polarization is to break the inversion symmetry of the crystal. Therefore, a particular attention is paid to so-called improper ferroelectricity, where this inversion symmetry is broken by some complex magnetic order. Such a situation typically occurs in frustrated magnets. In this section, we will argue that similar behavior is expected for the marcasite phase of NaO_2 .

Let us consider first the theoretical magnetic ground state, which was obtained from the analysis inter-molecular magnetic interactions without relativistic spin-orbit (SO) coupling. In this case, each sublattice forms the $\mathbf{Q} = (\pi/a, 0, \pi/c)$ structure, which is shown in Figure 7a. Strictly speaking, the relative directions of spins in two magnetic sublattices are determined by the SO interactions, as was argued in the previous section. However, such fine details of the magnetic structure are not important for the purposes of this section and here we neglect the effects of the SO coupling. The $\mathbf{Q} = (\pi/a, 0, \pi/c)$ structure can be transformed to itself by the inversion operation (\hat{I}) around any of the superoxide ions. Therefore, the inversion symmetry is preserved and the $\mathbf{Q} = (\pi/a, 0, \pi/c)$ structure is not ferroelectric. Similar situation occurs for $\mathbf{Q} = (0, \pi/b, \pi/c)$.

In the $\mathbf{Q} = (\pi/a, \pi/b, \pi/c)$ structure, each magnetic sub-

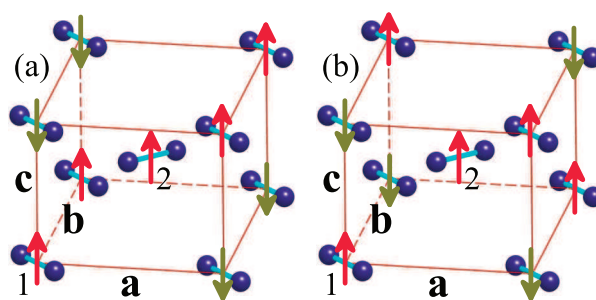


Fig. 7 (a) centrosymmetric $\mathbf{Q} = (\pi/a, 0, \pi/c)$ and (b) noncentrosymmetric $\mathbf{Q} = (\pi/a, \pi/b, \pi/c)$ magnetic structures in the marcasite phase of NaO_2 .

lattice can be transformed to itself by combining \hat{I} with the time-reversal operation (\hat{T}). This can be clearly seen in Figure 7b, by considering the transformations of ions of the sublattice 1 around the inversion center 2. However, the existence of the symmetry operation $\hat{T}\hat{I}$ would imply that the ion 2, which transforms to itself, is nonmagnetic. This contradicts to the electronic configuration of O_2^- , which has odd number of electrons. Then, the only possibility to reconcile the $\mathbf{Q} = (\pi/a, \pi/b, \pi/c)$ type of the magnetic ordering with the electronic configuration of O_2^- is to break the inversion symmetry. This will produce finite FE polarization \mathbf{P} . The mechanism is similar to the magnetic inversion symmetry breaking in the *E*-phase of manganites.⁴⁸ The polarization itself can be calculated using the Berry-phase theory,^{49,50} which was adopted for the effective Hubbard model.⁴⁸ Then, using the wavefunctions, obtained from the solution of the effective Hubbard model in the Hartree-Fock approximation, we obtain $\mathbf{P} = (53, 41, -248) \mu\text{C}/\text{m}^2$. Thus, the $\mathbf{Q} = (\pi/a, \pi/b, \pi/c)$ AFM order eventually breaks all symmetry operations (not only \hat{I}), and \mathbf{P} has finite values along all three orthorhombic axes. The absolute value of \mathbf{P} is more than order of magnitude smaller than in the *E*-phase of manganites.⁴⁸ Nevertheless, this is to be expected because the value of \mathbf{P} is controlled by the ratio of transfer integrals to the characteristic energy splitting between molecular π_g -orbitals (the atomic e_g orbitals in the case of manganites).⁵¹ The transfer integrals are smaller in superoxides, while the energy splitting is larger (due to larger Coulomb repulsion). Therefore, $|\mathbf{P}|$ should be smaller. Nevertheless, this value of \mathbf{P} is comparable or even larger than typical values of FE polarization in many transition-metal oxides with the deformed spin-spiral structure.⁴⁶

4 Conclusions

Using results of first-principles electronic structure calculations we have systematically studied the magnetic properties

of sodium superoxide. Our basic strategy was to construct a realistic model, describing the behavior of the magnetic degrees of freedom of NaO_2 in two crystallographic phases, depending on the local environment and relative orientation of the superoxide molecules. This model provides a transparent picture, explaining the magnetic properties of NaO_2 in different temperature regimes. Namely, the high-temperature pyrite phase is characterized by the orbital degeneracy and the disorder of the orbital degrees of freedom. This disorder naturally explains the weakly AFM character of inter-molecular magnetic interactions, that is seen in the experimental behavior of magnetic susceptibility. Moreover, we expect that such a behavior should be a generic property of alkali superoxides in the high-temperature state. The transition to the low-temperature marcasite phase of NaO_2 , which takes place around 196 K, is accompanied by the long-range order of the molecular π_g -orbitals. This orbital order naturally explains the magnetic properties of the marcasite phase, namely: (i) The character of isotropic exchange interactions, which is featured by the formation of the quasi-one-dimensional AFM $S = 1/2$ spin chains. This property is again reflected in the behavior of magnetic susceptibility in the marcasite phase, where χ rises with the temperature; (ii) Small values of the magnetic transition temperature, which is related to the one-dimensional character and frustration of isotropic exchange interactions; (iii) The behavior of anisotropic exchange interactions, which plays an important role in stabilizing the magnetic order between different magnetic sublattices. Moreover, we predict the multiferroic behavior of the marcasite phase. The theoretical value of polarization $|\mathbf{P}| \sim 250 \mu\text{C}/\text{m}^2$ is comparable with those observed in the transition-metal oxides, which are currently intensively studied in the context of multiferroic applications.

Thus, the alkali superoxides present an interesting example of magnetic materials on the basis of p -elements. They exhibit fascinating magnetic properties, including the spin-orbital coupled phenomena and the multiferroic effect. In this sense, the alkali superoxides can be regarded as molecular p -electron analogs of more traditional transition-metal oxides. The new aspect of alkali superoxides is that all these phenomena are realized on antibonding molecular orbitals of the superoxide complexes, that opens new functional possibilities for the design and control of the properties. Another important issue is the rotations of the superoxide molecules – the problem, which deserves a more fundamental theoretical analysis.

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